

Spin Crossover in Ferroperriclite in the Earth's Lower Mantle from $LDA + U_{sc}$ Calculations

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Abstract

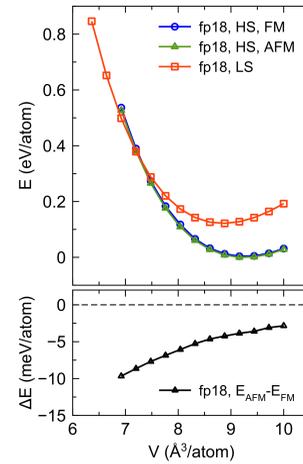
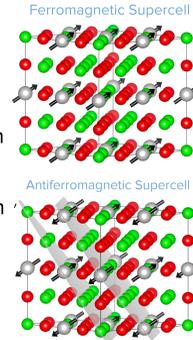
The high spin (HS) to low spin (LS) crossover of Fe^{2+} ions in ferroperriclite, $Mg_{(1-x)}Fe_xO$ (Fp), affects mantle properties such as density, elasticity, thermal properties, element partitioning, etc. It further affects the interpretation of lower mantle velocities. Here, we introduce a framework to compute energy with the inclusion of the non-ideal mixing energy term across spin-state changes using *ab initio* DFT+ U_{sc} calculations and compare results with available experimental and computational results. We explain how this framework differs from previous calculations that use the "ideal" HS-LS mixing model, and show the importance of constraining well the energy model and the computed properties from thermodynamic relationships of Fp.

Methods and calculation details

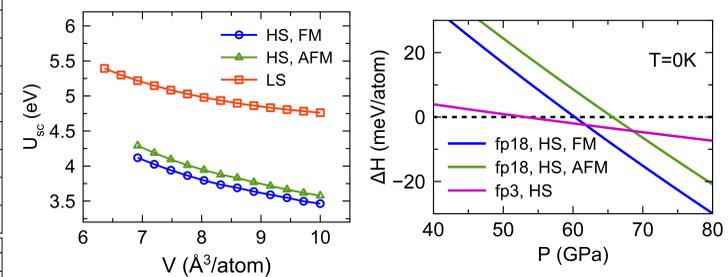
A supercell with 64 atoms for $x_{Fe} = 0.1875$

6 Fe, 26 Mg and 32 O atoms

- DFT+U [1]
- Self-consistent U_{sc} [2]
- Density Functional Perturbation Theory + U for phonons [3]
- LDA-PAW with spin polarization
- Quantum ESPRESSO [4]
- Phonopy code [5]
- $2 \times 2 \times 2$ k-point mesh
- qha to compute vibrational free energy [6]

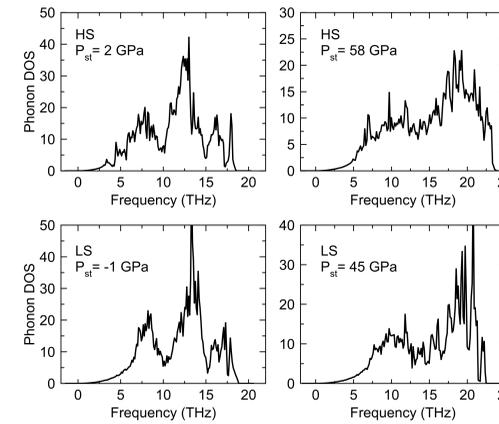


Energy, Hubbard parameters and Enthalpy



Left: E vs. V for the three states and energy difference between the HS-AFM and HS-FM. Middle: self-consistent Hubbard parameters U for different spin states. Right: enthalpy difference between HS and LS.

Vibrational DOS for HS and LS



Ideal solid solution of HS and LS ferroperriclite

Low spin fraction $n = n_{LS} / (n_{LS} + n_{HS})$

Ideal mixing volume $V = (1 - n)V_{HS} + nV_{LS}$

Ideal mixing gibbs free energy

$$G_{ideal} = (1 - n)G_{HS} + nG_{LS} + G_{mix}$$

High/low spin Gibbs free energy $G_{HS/LS} = F_{HS/LS} + PV_{HS/LS}$

$$G_{HS/LS} = F_{HS/LS}^{stat+vib} + G_{HS/LS}^{mag} + PV_{HS/LS}$$

Magnetic contribution

$$G_{mag} = -TS_{mag} = -k_B T x_{Fe} (1 - n) \ln[m(2S + 1)]$$

• x_{Fe} is the iron concentration

• S is iron spin quantum number

$S = 2$ for HS and $S = 0$ for LS

• m is electronic configuration degeneracy

$m = 3$ for HS and $m = 1$ for LS

Ideal energy of mixing

$$G_{mix} = -TS_{conf} = k_B T x_{Fe} [n \ln n + (1 - n) \ln(1 - n)]$$

n is calculated by minimizing G_{ideal}

$$f(P, T, n) = \Delta G_{LS-HS} + k_B T x_{Fe} \ln \left[\frac{n}{1-n} (m(2S + 1)) \right] = 0$$

which gives equilibrium $n(P, T)$:

$$n = \frac{1}{1 + m(2S + 1) \exp \left[\frac{\Delta G_{LS-HS}}{k_B T x_{Fe}} \right]}$$

where $\Delta G_{LS-HS} = G_{LS} - G_{HS}$

Non-ideal solid solution

Non-ideal gibbs free energy

$$G_{non-ideal} = G_{ideal} + G_{ex}$$

assume the excess energy is dominated by its static part

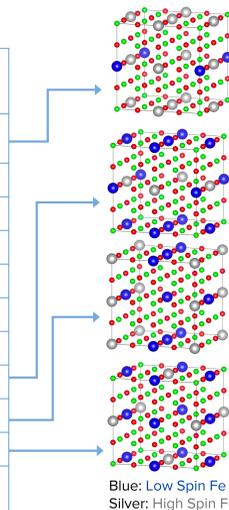
$$G_{ex}(P, T, n) \approx H_{ex, st}(P, n)$$

n is calculated by minimizing $G_{non-ideal}$. Here, we would like to numerically solve

$$f(P, T, n) = \Delta G_{LS-HS} + k_B T x_{Fe} \ln \left[\frac{n}{1-n} (m(2S + 1)) \right] + \frac{\partial H_{ex}}{\partial n} = 0$$

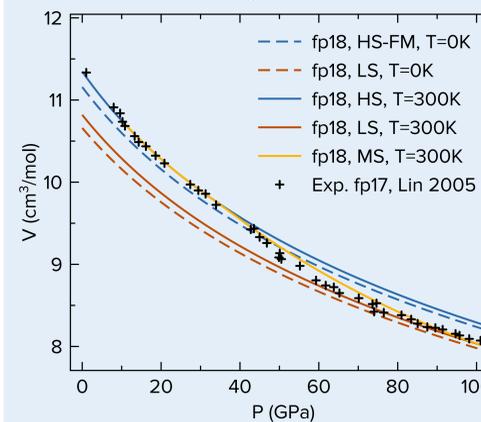
Unique configurations

Low spin fraction n	Space group of unique configurations	Number of equivalent configurations
$\frac{1}{6}$	#123 P4/mmm	1
$\frac{1}{3}$	#123 P4/mmm	6
	#131 P4_2/mmc	6
	#139 I4/mmm	3
$\frac{1}{2}$	#123 P4/mmm	6
	#221 Pm-3m	2
$\frac{2}{3}$	#47 Pmmm	12
	#123 P4/mmm	6
	#131 P4_2/mmc	6
$\frac{5}{6}$	#139 I4/mmm	3
	#123 P4/mmm	1



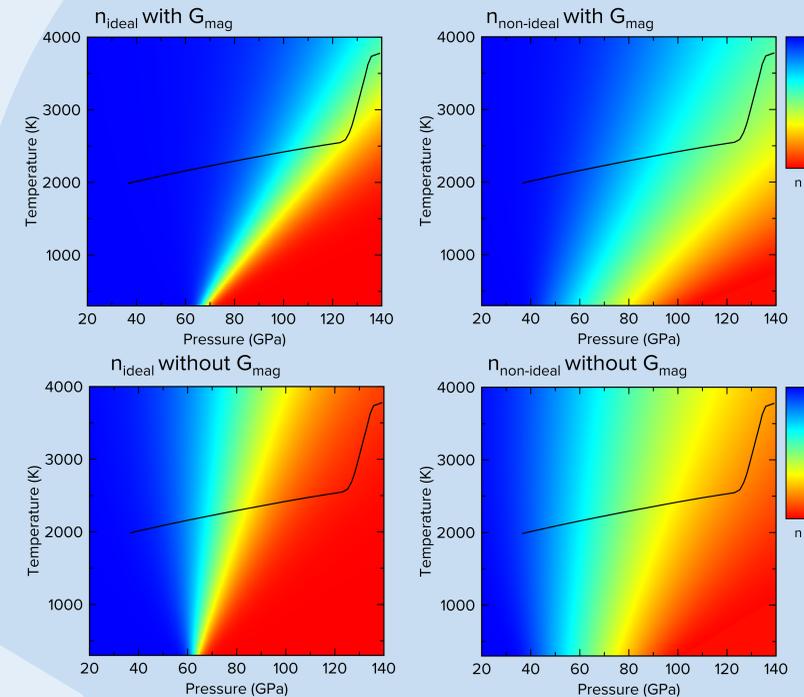
Equation of state at 300 K

Compression curve of fp18. The experimental data is at 300 K with $x_{Fe}=17\%$ ferroperriclite [7].



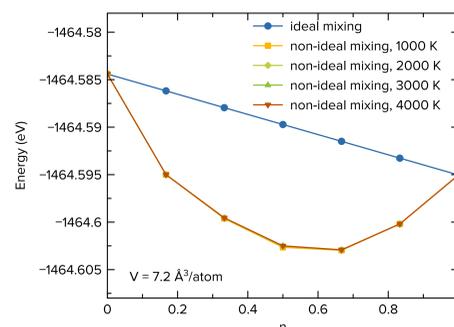
Spin-Crossover LS fraction n

Temperature-dependent spin-crossover ranges with $x_{Fe}=18.75\%$ based on the different models as labelled. The black line indicates the geotherm [8, 9].



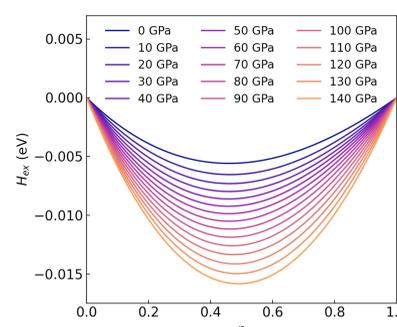
Non-ideal Mixing – Methods

Compute the MS static energy through the weighted average using Boltzmann factor of the unique configurations at each n



Use polynomial 3rd order to fit $H_{ex}(n)$

- $H_{ex}(n=0) = 0$ and $H_{ex}(n=1) = 0$
- $H_{ex}(n) = a n^3 + b n^2 - (a+b) n$



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Acknowledgements

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