

Influence of nanoCaCO $_3$ seeding on the hydration and carbonation behavior of MgO based cements

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Background & Motivation

Reject brine from desalination plants is typically discharged into the ocean, posing environmental concerns by having concentrations of ions five times greater than seawater as well as other harmful constituents from treatment. It is possible to extract resources of interest that can be processed into valuable resources. Mg(OH), is extracted from brine via addition of NaOH and can be calcinated at ~500°C to produce reactive MgO. This material is used in the production of reactive MgO cements (RMC) that have the potential of replacing industry standard Ordinary Portland Cement (OPC), given its comparable properties and more sustainable production. CaCO₂ can also be precipitated from brine through the addition of NaOH and bubbling of CO₂. This has potential to be used as a nucleation seed in cement which can promote hydration and carbonation and consequently enhance its properties.

RMCs can harden and gain strength through CO₂ curing, where the material "soaks up" CO2 to form solid carbonates. Under ambient conditions MgO reacts with water, i.e. undergoes hydration, to produce Mg(OH), which leads to setting. Under carbonation, the Mg(OH), reacts with CO, and forms hydrated magnesium carbonates (HMC), which densifies and cements the structure to lead to strength gain.

Some materials used as nucleation seeds like hydromagnesite have been proven to enhance RMCs properties¹. Others like nanoCaCO, have only been shown to work with OPC². As we can extract CaCO₂ and MgO from reject brine, this project is focused on the influence of nanoCaCO, seeding on the hydration and carbonation of RMCs.

Materials & Procedures

- I Materials:
 - Water
 - Sand (aggregate)
 - MgO (binder)
 - nanoCaCO₂ (seed)
- Mix Design:
 - 1. Control
 - 2.5% nanoCaCO₃ replacement by weight
- Samples:
 - 2in cubes (x2 for each)
 - Air curing (0-24h, 25°C, room conditions)
 - Carbon curing (Day 1-28, 20% CO2, 25°C, 80% RH)



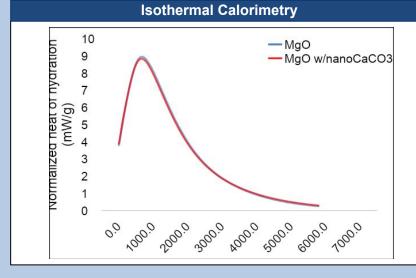




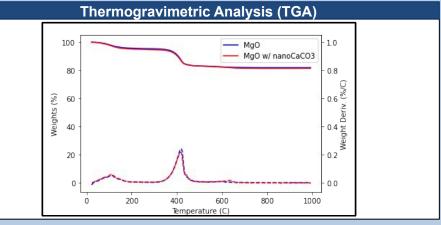
Testina

We can measure and compare the degree of hydration and carbonation through three tests to determine the influence of nanoCaCO₂ seeding.

- Isothermal Calorimetry: Measure of heat flow over 72h, indicator of kinetics and degree of hydration
- Strength testing: Indirect measure and direct assessment of the effectiveness of carbon curing. Representative strength taken from sample average
- Thermogravimetric analysis (TGA): Measures thermal decomposition of phases through mass loss, potential indicator of degree of carbonation via quantifying the amount of carbonate phases



Strength Testing Compressive strength (MPa) MgO MaO (w/nanoCaCO3)



In this project, we set out to investigate the influence of nanoCaCO₂ seeds on the behavior of reactive magnesium cement. We expect seeding to improve the kinetics, degree of hydration and carbonation of RMCs, leading to enhanced strength. Our results suggest that nanoCaCO₂ is not a good seed for this type of cement. The isothermal calorimetry shows very little difference between the two samples, meaning that the nanoCaCO₂ had little to no impact on the degree of hydration or the kinetics. In terms of strength, plain MgO had a representative strength that was 18.2% higher than MgO w/ nanoCaCO₂. Finally, there was also little difference in the TGA curves that show similar degree of hydration and carbonation. The peak at ~100C corresponds to water and HMC, the peak at 400 C is for brucite ad the peak at >500 C is for HMC. This suggests that plain MgO had a higher degree of hydration but slightly lower carbonation.

In the end, using nanoCaCO, in magnesium based cements does not improve its resulting properties. Clearly, we would need to further our research to fully determine the usefulness of this seed. Perhaps by increasing CO₂ content or increasing replacement amounts of nanoCaCO, can produce a cement comparable to OPC.

1 Unluer, C., & Al-Tabbaa, A. (2013). Impact of hydrated magnesium carbonate additives on the carbonation of reactive MgO cements, Cement and Concrete Research, 54, 87-97. 2 Sato, T., & Beaudoin, J. J. (2019). Effect of nano-CaCO3 on hydration of cement containing supplementary cementitious materials. In ICE Themes Low Carbon Concrete (pp. 231-248). ICE Publishing. 3 Ma, S., Akca, A. H., Esposito, D., & Kawashima, S. (2020). Influence of aqueous carbonate species on hydration and carbonation of reactive MgO cement. Journal of CO2 Utilization, 41, 101260.

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Mass ratios

MgO

0.95

Water Sand

2

Aix 1 0.9

0.9

nanoCaCO3

0.05



Conclusion & Discussion

References

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